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## Structure Reports

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## Key indicators

Single-crystal X-ray study
$T=295 \mathrm{~K}$
Mean $\sigma(\mathrm{C}-\mathrm{C})=0.004 \AA$
H -atom completeness $81 \%$
Disorder in solvent or counterion
$R$ factor $=0.028$
$w R$ factor $=0.067$
Data-to-parameter ratio $=15.4$

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.
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## catena-Poly[[[triaquacadmium(II)]- $\mu$-3-carboxylato-phenoxyacetato- $\left.\kappa^{3} O^{3}, O^{3^{\prime}}: O^{\prime \prime}\right]$ sesquihydrate]

The dicarboxylate ligand in the title compound, $\left[\mathrm{Cd}\left(\mathrm{C}_{9} \mathrm{H}_{6} \mathrm{O}_{5}\right)\right.$ $\left.\left(\mathrm{H}_{2} \mathrm{O}\right)_{3}\right] \cdot 1.5 \mathrm{H}_{2} \mathrm{O}$, bridges triaquacadmium groups into a linear chain; the carboxylate arm that is directly connected to the aromatic ring behaves as a chelate, whereas the carboxylate arm of the oxyacetate portion binds in a monodentate mode. The octahedral environment of cadmium is distorted towards a pentagonal bipyramid, owing to a long $\mathrm{Cd} \cdots \mathrm{O}$ interaction of 2.721 (2) $\AA$.

## Comment

A recent study of the cadmium derivative of 4-carboxyphenoxyacetic acid describes the isolation of a triaqua compound whose seven-coordinate metal exists in a penta-gonal-bipyramidal environment (Gao et al., 2005). Both carboxylate arms of the dicarboxylate function in the chelating mode, with the four $\mathrm{Cd}-\mathrm{O}$ bond distances ranging from 2.356 (2) to 2.444 (2) $\AA$. The corresponding 3-carboxyphenoxyacetic acid also affords a triaquacadmium compound. However, probably owing to the greater flexibility of the ligand, the packing is less efficient, so that the compound crystallizes with one-and-a-half uncoordinated water molecules, (I) (Fig. 1). Compound (I) exists as a linear chain that propagates along the longest axis of the unit cell. Of the two carboxylate arms, that connected to the aromatic ring functions as a chelate, whereas that belonging to the oxyacetate portion is only monodentate. However, a long-range interaction between Cd1 and O5 is sufficiently close [2.721 (2) A] to distort the octahedral geometry to a pentagonal bipyramid. The chains are linked by hydrogen bonds (Table 2) into a three-dimensional network.


## Experimental

Cadmium dinitrate tetrahydrate ( $0.62 \mathrm{~g}, 2 \mathrm{mmol}$ ) and 3-carboxyphenoxyacetic acid ( $0.39 \mathrm{~g}, 2 \mathrm{mmol}$ ) were dissolved in a small volume of water ( 30 ml ); $0.2 M$ sodium hydroxide was added in drops to the solution until it showed a pH of 6 . The solution was set aside for the formation of colourless prisms; crystallization took several weeks. CH analysis, calculated for $\mathrm{C}_{9} \mathrm{H}_{15} \mathrm{O}_{9.5} \mathrm{Cd}$ : C $27.89, \mathrm{H} \mathrm{3.90} \mathrm{\% ;} \mathrm{found:}$ C 27.16, H 4.03\%.

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## Crystal data

$\left[\mathrm{Cd}\left(\mathrm{C}_{9} \mathrm{H}_{6} \mathrm{O}_{5}\right)\left(\mathrm{H}_{2} \mathrm{O}\right)_{3}\right] \cdot 1.5 \mathrm{H}_{2} \mathrm{O}$
$M_{r}=387.61$
Triclinic, $P \overline{1}$
$a=7.029$ (1) $\AA$
$b=8.829$ (2) $\AA$
$c=11.720(2) \AA$
$\alpha=80.62(3)^{\circ}$
$\beta=72.66(3)^{\circ}$
$\gamma=76.52(3)^{\circ}$
$V=671.7(2) \AA^{3}$

## Data collection

Rigaku RAXIS-RAPID IP diffractometer

## $\omega$ scans

Absorption correction: multi-scan (ABSCOR; Higashi, 1995)
$T_{\text {min }}=0.460, T_{\text {max }}=0.753$
5896 measured reflections

## Refinement

Refinement on $F^{2}$

$$
\begin{aligned}
& w=1 /\left[\sigma^{2}\left(F_{\mathrm{o}}{ }^{2}\right)+(0.0358 P)^{2}\right. \\
& +0.544 P \text { ] } \\
& \text { where } P=\left(F_{\mathrm{o}}{ }^{2}+2 F_{\mathrm{c}}{ }^{2}\right) / 3 \\
& (\Delta / \sigma)_{\max }=0.001 \\
& \Delta \rho_{\max }=0.71 \mathrm{e}^{-3} \\
& \Delta \rho_{\min }=-0.57 \mathrm{e}^{-3}
\end{aligned}
$$

## $Z=2$

$D_{x}=1.916 \mathrm{Mg} \mathrm{m}^{-3}$
Mo $K \alpha$ radiation
Cell parameters from 5430 reflections
$\theta=3.1-27.5^{\circ}$
$\mu=1.67 \mathrm{~mm}^{-1}$
$T=295$ (2) K
Block, colourless
$0.36 \times 0.25 \times 0.18 \mathrm{~mm}$

2969 independent reflections
2671 reflections with $I>2 \sigma(I)$
$R_{\text {int }}=0.021$
$\theta_{\text {max }}=27.5^{\circ}$
$h=-9 \rightarrow 9$
$k=-11 \rightarrow 11$
$l=-15 \rightarrow 14$
$R\left[F^{2}>2 \sigma\left(F^{2}\right)\right]=0.028$
$w R\left(F^{2}\right)=0.067$
$S=1.05$
2969 reflections
193 parameters
H -atom parameters constrained


Figure 1
A plot of a portion of the chain structure of (I). The disordered uncoordinated water molecules are not shown. Displacement ellipsoids are drawn at the $50 \%$ probability level, and H atoms are drawn as spheres of arbitrary radii. [Symmetry code: (i) $x, y, 1+z$.]

The carbon-bound H atoms were positioned geometrically $[\mathrm{C}-\mathrm{H}$ 0.93 or $0.97 \AA$ ] and were included in the refinement in the ridingmodel approximation, with $U_{\text {iso }}(\mathrm{H})=1.2 U_{\text {eq }}(\mathrm{C})$. The H atoms of the coordinated water molecules were similarly treated $[\mathrm{O}-\mathrm{H} 0.85 \AA$, $\left.U_{\text {iso }}(\mathrm{H})=1.5 U_{\text {eq }}(\mathrm{O})\right]$; these were rotated to fit the electron density.

Of the two independent uncoordinated water molecules, one is disordered over two positions ( $\mathrm{O} 4 w$ and $\mathrm{O} 4 w^{\prime}$ ), while the other (O5w) is disordered over an inversion centre. A consideration of possible hydrogen-bonding interactions led to the assignment of half occupancies for these atoms. H atoms could not be placed in any chemically sensible positions owing to the disorder. Atom O4w is $3 \AA$ from $\mathrm{O} 5 w^{\text {ii }}$ while $\mathrm{O} 4 w^{\prime}$ is $2.7 \AA$ from $\mathrm{O} 5 w$ [symmetry code: (ii) $3-x$, $2-y, 1-z]$.

Data collection: RAPID-AUTO (Rigaku, 1998); cell refinement: RAPID-AUTO; data reduction: CrystalStructure (Rigaku/MSC, 2002); program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: ORTEPII (Johnson, 1976); software used to prepare material for publication: SHELXL97.

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## References

Gao, S., Liu, J.-W., Huo, L.-H. \& Zhao, J.-G. (2005). Acta Cryst. E61, m1092m1094.
Higashi, T. (1995). ABSCOR. Rigaku Corporation, Tokyo, Japan.
Johnson, C. K. (1976). ORTEPII. Report ORNL-5138. Oak Ridge National Laboratory, Tennessee, USA.
Rigaku (1998). RAPID-AUTO. Rigaku Corporation, Tokyo, Japan.
Rigaku/MSC (2002). CrystalStructure. Rigaku/MSC Inc., 9009 New Trails Drive, The Woodlands, TX 77381-5209, USA.
Sheldrick, G. M. (1997). SHELXL97 and SHELXS97. University of Göttingen, Germany.

