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Key indicators

Single-crystal X-ray study T = 295 KMean $\sigma(C-C) = 0.004 \text{ Å}$ H-atom completeness 81% Disorder in solvent or counterion R factor = 0.028 wR factor = 0.067 Data-to-parameter ratio = 15.4

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catena-Poly[[[triaquacadmium(II)]- μ -3-carboxylato-phenoxyacetato- $\kappa^3 O^3, O^{3'}: O''$] sesquihydrate]

The dicarboxylate ligand in the title compound, $[Cd(C_9H_6O_5)-(H_2O)_3]\cdot 1.5H_2O$, bridges triaquacadmium groups into a linear chain; the carboxylate arm that is directly connected to the aromatic ring behaves as a chelate, whereas the carboxylate arm of the oxyacetate portion binds in a monodentate mode. The octahedral environment of cadmium is distorted towards a pentagonal bipyramid, owing to a long Cd···O interaction of 2.721 (2) Å.

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Comment

A recent study of the cadmium derivative of 4-carboxyphenoxyacetic acid describes the isolation of a triaqua compound whose seven-coordinate metal exists in a pentagonal-bipyramidal environment (Gao et al., 2005). Both carboxylate arms of the dicarboxylate function in the chelating mode, with the four Cd-O bond distances ranging from 2.356 (2) to 2.444 (2) Å. The corresponding 3-carboxyphenoxyacetic acid also affords a triaquacadmium compound. However, probably owing to the greater flexibility of the ligand, the packing is less efficient, so that the compound crystallizes with one-and-a-half uncoordinated water molecules, (I) (Fig. 1). Compound (I) exists as a linear chain that propagates along the longest axis of the unit cell. Of the two carboxylate arms, that connected to the aromatic ring functions as a chelate, whereas that belonging to the oxyacetate portion is only monodentate. However, a long-range interaction between Cd1 and O5 is sufficiently close [2.721 (2) Å] to distort the octahedral geometry to a pentagonal bipyramid. The chains are linked by hydrogen bonds (Table 2) into a three-dimensional network.



Experimental

Cadmium dinitrate tetrahydrate (0.62 g, 2 mmol) and 3-carboxyphenoxyacetic acid (0.39 g, 2 mmol) were dissolved in a small volume of water (30 ml); 0.2 *M* sodium hydroxide was added in drops to the solution until it showed a pH of 6. The solution was set aside for the formation of colourless prisms; crystallization took several weeks. CH analysis, calculated for $C_9H_{15}O_{9.5}Cd$: C 27.89, H 3.90%; found: C 27.16, H 4.03%.

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metal-organic papers

Z = 2

 $D_x = 1.916 \text{ Mg m}^{-3}$

Cell parameters from 5430

Mo $K\alpha$ radiation

reflections

 $\theta = 3.1 - 27.5^{\circ}$ $\mu=1.67~\mathrm{mm}^{-1}$

T = 295 (2) K

 $R_{\rm int} = 0.021$

 $\theta_{\rm max} = 27.5^{\circ}$

 $h = -9 \rightarrow 9$

 $k = -11 \rightarrow 11$

 $l = -15 \rightarrow 14$

Block, colourless

 $0.36 \times 0.25 \times 0.18 \text{ mm}$

2969 independent reflections

2671 reflections with $I > 2\sigma(I)$

Crystal data

[Cd(C₉H₆O₅)(H₂O)₃]·1.5H₂O $M_r = 387.61$ Triclinic, $P\overline{1}$ a = 7.029 (1) Å b = 8.829 (2) Å c = 11.720 (2) Å $\alpha = 80.62$ (3) $\beta = 72.66 (3)^{\circ}$ $\gamma = 76.52$ (3) V = 671.7 (2) Å³

Data collection

Rigaku RAXIS-RAPID IP diffractometer ω scans Absorption correction: multi-scan (ABSCOR; Higashi, 1995) $T_{\min} = 0.460, \ \tilde{T}_{\max} = 0.753$ 5896 measured reflections

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_0^2) + (0.0358P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.028$	+ 0.544P]
$wR(F^2) = 0.067$	where $P = (F_0^2 + 2F_c^2)/3$
S = 1.05	$(\Delta/\sigma)_{\rm max} = 0.001$
2969 reflections	$\Delta \rho_{\rm max} = 0.71 \text{ e } \text{\AA}^{-3}$
193 parameters	$\Delta \rho_{\rm min} = -0.57 \text{ e} \text{ \AA}^{-3}$
H-atom parameters constrained	

Table 1

Selected geometric parameters (Å, °).

Cd1-O1	2.445 (2)	Cd1-O1w	2.337 (3)
Cd1-O2	2.301 (2)	Cd1 - O2w	2.268 (2)
Cd1-O4 ⁱ	2.281 (2)	Cd1 - O3w	2.399 (2)
O1-Cd1-O2	55.0 (1)	O2-Cd1-O3w	95.7 (1)
$O1-Cd1-O4^{i}$	144.6 (1)	O4 ⁱ -Cd1-O1w	92.8 (1)
O1-Cd1-O1w	97.7 (1)	$O4^i - Cd1 - O2w$	131.2 (1)
O1-Cd1-O2w	83.8 (1)	$O4^i - Cd1 - O3w$	88.2 (1)
O1-Cd1-O3w	87.2 (1)	O1w-Cd1-O2w	83.1 (1)
$O2-Cd1-O4^{i}$	90.6 (1)	O1w-Cd1-O3w	169.9 (1)
O2-Cd1-O1w	94.3 (1)	O2w-Cd1-O3w	88.7 (1)
O2-Cd1-O2w	138.2 (1)		

Symmetry code: (i) x, y, z + 1.

Table 2

Hydrogen-bond geometry (Å, °).

$D-\mathrm{H}\cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdot \cdot \cdot A$
$O1w - H1w1 \cdots O4w$	0.85	2.27	3.015 (8)	147
$O1w - H1w1 \cdots O4w'$	0.85	1.83	2.593 (8)	148
$O1w - H1w2 \cdots O4^{ii}$	0.85	1.95	2.732 (3)	153
$O2w - H2w1 \cdots O3w^{iii}$	0.85	2.09	2.841 (3)	147
$O2w - H2w2 \cdots O5^{iv}$	0.85	2.02	2.821 (4)	157
$O3w - H3w2 \cdots O1w^{v}$	0.85	2.21	2.959 (3)	147
$O3w - H3w1 \cdots O2^{vi}$	0.85	1.84	2.684 (3)	175

Symmetry codes: (ii) -x + 2, -y + 1, -z + 1; (iii) -x + 1, -y + 2, -z + 2; (iv) -x + 1, -y + 2, -z + 1; (v) x - 1, y, z; (vi) -x + 1, -y + 1, -z + 2.



Figure 1

A plot of a portion of the chain structure of (I). The disordered uncoordinated water molecules are not shown. Displacement ellipsoids are drawn at the 50% probability level, and H atoms are drawn as spheres of arbitrary radii. [Symmetry code: (i) x, y, 1 + z.]

The carbon-bound H atoms were positioned geometrically [C-H 0.93 or 0.97 Å] and were included in the refinement in the ridingmodel approximation, with $U_{iso}(H) = 1.2U_{eq}(C)$. The H atoms of the coordinated water molecules were similarly treated [O-H 0.85 Å, $U_{iso}(H) = 1.5U_{eq}(O)$; these were rotated to fit the electron density.

Of the two independent uncoordinated water molecules, one is disordered over two positions (O4w and O4w'), while the other (O5w) is disordered over an inversion centre. A consideration of possible hydrogen-bonding interactions led to the assignment of half occupancies for these atoms. H atoms could not be placed in any chemically sensible positions owing to the disorder. Atom O4w is 3 Å from $O5w^{ii}$ while O4w' is 2.7 Å from O5w [symmetry code: (ii) 3 - x, 2 - y, 1 - z].

Data collection: RAPID-AUTO (Rigaku, 1998); cell refinement: RAPID-AUTO: data reduction: CrvstalStructure (Rigaku/MSC, 2002); program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: ORTEPII (Johnson, 1976); software used to prepare material for publication: SHELXL97.

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